

**UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**GCE Advanced Subsidiary Level and GCE Advanced Level**

**MARK SCHEME for the October/November 2010 question paper  
for the guidance of teachers**

**9701 CHEMISTRY**

**9701/53**

Paper 5 (Planning, Analysis and Evaluation),  
maximum raw mark 30

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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Question	Sections	Indicative material	Mark
<b>(f)</b>	PLAN method	<p>There are 6 marking points which are,</p> <p>Three columns, concentration of hydrogen peroxide, time and rate (or 1/time). Ignore other columns</p> <p>Each column needs a correct unit in correct format i.e. /mol dm<sup>-3</sup>, /s, s<sup>-1</sup> or the use of brackets (s). Accept seconds, minutes (not sec, min or m), or not M or molarity.</p> <p>Two marks for 5 or 6 correct points. One mark for 3 or 4 correct points. No marks for 2 or less correct points.</p>	[2]
	<b>Total</b>		<b>[15]</b>

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Question	Sections	Indicative material	Mark
<b>2 (a)</b>	ACE Data	One of the two column headings correct in heading, unit and expression.	[1]
		The calculations are correct in both columns (first two and last two in each) and both columns are fully completed (to 3SF). (One mark for each column).	[2]
		If an expression is not given and all the data is totally correct then the last 2 marks are available.	
		ECF data from incorrect expressions provided correctly calculated (and provided some attempt at a titration calculation is made). For incorrect expressions check calculate test data. Then the last 2 marks are available.	
<b>(b)</b>	ACE Data	Give one mark for unambiguously labelling and scaling the x-axis and the y-axis provided the plotted points cover at least half the scalings in both directions. Plot may be either way round. Headings could be names or D or E	[1]
		Give one mark for correctly plotting the first, last and anomalous points and those that deviate significantly from the line ( $\pm \frac{1}{2}$ square except where a grid line is involved). All 10 data points must be plotted on the grid. ECF plots of incorrect data.	[1]
		Give one mark for drawing a 'line of best fit' which must pass through the origin ( $\pm \frac{1}{2}$ ).	[1]
<b>(c)</b>	ACE Data	The anomalies must be ringed and normally must include the two points furthest away from the drawn line on each side of the line (ignore other anomalies). If all the anomalies are on one side of the line – ring the furthest away (also ignore other anomalies). Accept only one anomaly if that is the situation where there is only one anomaly (the candidate may not have ringed all the anomalies). This mark negated if more than 5 anomalies.	[1]
		For each of the two different anomalies an appropriate explanation gains one mark. Explanations must be related to the particular point and include the nature of the deviation.  Award 1 mark for two correct explanations not properly linked to a point.	[2]

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Question	Sections	Indicative material	Mark	
<b>(d)</b>	<b>(i)</b>	ACE data	For appropriately drawn lines on the graph with correctly deduced intercepts (+/- ½ square except where a grid line is involved) give one mark.	[1]
		<b>(ii)</b>	Correctly calculates the value of the gradient. This should be in the order of 16.3/0.061. ECF incorrect intercepts.	[1]
			Yes, Since the results produce a good linear/straight line graph, the procedure is OK. Normally a “no” answer is not acceptable. Do not accept an unjustified “yes” answer.	[1]
<b>(e)</b>	ACE Data	Any facilitation that takes the (succinic) acid into an aqueous phase will suffice e.g. to ensure that all the reactants are mixed in the aqueous layer / so the reactants are in solution in water for the neutralisation / makes the titration work because the acid is in the aqueous layer / extract or mix the acid into the aqueous phase to react / produce H <sup>+</sup> <sub>(aq)</sub> to react with the alkali. Answers must have the acid in an aqueous phase.	[1]	
<b>(f)</b>	ACE Data	Low (titre) values. Thus % errors are high (consequential). Percentage error is required.	[1] [1]	
	<b>Total</b>		<b>[15]</b>	